PHARMACOPOEIAL DISCUSSION GROUP

SIGN-OFF DOCUMENT
CODE: G-07
NAME: Elemental Impurities

It is understood that sign-off covers the technical content of the draft and each party will adapt it as necessary to conform to the usual presentation of the pharmacopoeia in question; such adaptation includes stipulation of the particular pharmacopoeia's reference materials and general chapters.

Harmonised provisions:

Provision	EP	IPC	JP	USP
Introduction	+	+	+	+
Analytical Procedures 1 and 2	+	+	+	+
Requirements for Procedure Validation	+	+	+	+
Procedures for Limit Tests	+	+	+	+
Procedures for Quantitative Tests	+	+	+	+
Glossary	+	+	+	+

Legend
+: will adopt and implement
–: will not stipulate

Non-harmonised provisions:

None.

Local requirements

EP	IPC	JP	USP
Sample preparation: • The sentence on safety considerations when using concentrated acids is omitted. • Addition of a section on labware selection. *Procedure and Detection Technique* • Inclusion of references to Ph. Eur. general chapters *Analytical procedures 1 and 2:*	Detail about Elemental impurities -limits use of "test solution" instead of "sample solution" and "reference solution" instead of "standard solution" *Introduction*: • The note to clarify analytical methods other than the methods described in this chapter can be used if validated.	*Introduction*: • The sentence about the purpose of this chapter. • The note to clarify analytical methods other than the methods described in this chapter can be used if validated. *Analytical Procedures 1 and 2*: • "if necessary" will be added to "Sample stock solution" to clarify it is not necessary to add stabilizer	Addition of *Speciation* section *Analytical Procedures 1 and 2:* • "Standard solution 1" and "Standard solution 2" changed to "Standardization" solution 1" and "Standardization solution 2" *Glossary:* • Definition of "Concentrated acid" to include Aqua regia

• Use of "Calibration instead of "*Standardization*" • Use of "test solution" instead of "sample solution" *Requirements for procedure validation:* • Addition of a sentence connecting with other Ph. Eur. general chapters *Glossary*: • Addition of a sentence on the availability of reference material.	Heading "Sample preparation" changed to "Test solution preparation" Indirect *solution*: Added "Note- The test solution preparation scheme should yield sufficient sample to allow quantification of each element at the limit specified in the corresponding monograph or chapter" *Analytical procedures 1 and 2:* • Inclusion of references to IP general chapters • "Standard solution 1" and "Standard solution 2" changed to "Reference solution (a)" and "reference solution (b)" • "Sample stock solution" and "sample solution" changed to "Test solution (a)" and "Test solution (b)" respectively' *Requirements for procedure validation:* • Addition of sentence connecting with other IP general chapters Glossary: Definition of "*Cross validation*,"	depending on the matrix. • "Generally," will be added to "Rinse" to clarify other acids can be used for rinse if "memory effect" is observed on the apparatus. *Requirements for Procedure Validation*: • The sentence to clarify the validation method and criteria may be changed depending on the content level of elemental impurities. • The sentence to explain about the difference between the JP existing general test <2.63> on ICP and this chapter. *Procedures for Quantitative Tests;* • The supplementary information on the preparation procedure for standard solutions and test samples. • The acceptance criterion of the Quantification Limit will be replaced by "The QL is smaller or equal to 50% of *Target concentration*." *Glossary* ; • Different wording for "*Target elements*" and "*Target limit or Target concentration*" • Definition of "*Cross validation*,"	• Addition of definition of "Aqua regia" *Appendix:* • Appropriate reference materials to include example of an NMI

European Pharmacopoeia

Signature	Name	Date
DocuSigned by: Cathie VIELLE 5D202E6E19D1466...	C. Vielle	17 June 2024

Indian Pharmacopoeia Commission

Signature	Name	Date
DocuSigned by: 6CD705E38A2848B...	Gaurav Pratap Singh	19 June 2024

Japanese Pharmacopoeia

Signature	Name	Date
DocuSigned by: Yoshiro Saito for K. Nakai 9BF72DA462C9442...	Yoshiro Saito	18 June, 2024

United States Pharmacopeia

Signature	Name	Date
DocuSigned by: Kevin Moore A7467E52FCC94E9...	Kevin Moore	6/13/2024

ELEMENTAL IMPURITIES

INTRODUCTION

This chapter describes two analytical procedures (Procedures 1 and 2) and validation criteria for the evaluation of the levels of elemental impurities. The chapter permits the use of any procedure that meets the validation criteria specified in this chapter. As the chemical composition of the considered substances and the specification limits for the element(s) of interest vary considerably, it is difficult to describe all suitable sample preparation and measurement methods. By means of validation studies, analysts will confirm that the analytical procedure is suitable for use on specified material. It is not necessary to verify whether or not the same result can be obtained from the corresponding analyses for the same sample against either procedure 1 or 2.

As elemental impurities may be ubiquitous, they have the potential to be present in trace amounts therefore special precautions may be necessary to avoid sample contamination.

Sample Preparation

Forms of sample preparation include *Neat*, *Direct aqueous solution*, *Direct organic solution*, and *Indirect solution*. The selection of the appropriate sample preparation depends on the material under test and is the responsibility of the analyst. When a sample preparation is not indicated in the monograph, an analyst may use any appropriately validated sample preparation procedure, including but not limited to procedures described below. In cases where spiking of a material under test is necessary to provide an acceptable signal intensity, the blank should be spiked with the same *Target elements*, and where possible, using the same spiking solution. The material or mixture under test must be spiked before any sample preparation steps are performed. Standard solutions may contain multiple *Target elements*. [Note: if intended for a quantitative test, appropriate material handling procedures should be followed e.g. volatile liquids should be pipetted, viscous liquids should be weighed.]

Neat: Used for liquids or analytical procedures that allow the examination of unsolvated samples.
Direct aqueous solution: Used when the sample is soluble in an aqueous solvent.
Direct organic solution: Used when the sample is soluble in an organic solvent.
Indirect solution: Generally, an indirect solution is obtained when a material is not directly soluble in aqueous or organic solvents. Total digestion is the preferred sample preparation approach to obtain an *indirect solution*. Digest the sample using the *Closed vessel digestion* procedure provided below or one similar to it.
Closed vessel digestion: This sample preparation procedure is designed for samples that must be digested in a *Concentrated acid* using a closed vessel digestion apparatus. *Closed vessel digestion* minimizes the loss of volatile impurities. The choice of a *Concentrated acid* depends on the sample matrix. The use of any of the Concentrated acids may be appropriate, but each introduces inherent safety risks. Therefore, appropriate safety precautions should be used at all times. [Note—Weights and volumes provided may be adjusted to meet the requirements of the digestion apparatus used.]
An example procedure that has been shown to have broad applicability is the following. Dehydrate and predigest 0.5 g of material under test in 5 mL of freshly prepared *Concentrated acid*. Allow to sit loosely covered for 30 min in a fume hood. Add an additional 10 mL of *Concentrated acid*, and digest, using a closed vessel technique, until digestion or extraction results in a clear solution. Repeat, if necessary, by adding an additional 5 mL of *Concentrated acid*. [Note—Where closed vessel digestion is necessary, follow the manufacturer's recommended procedures to ensure safe use.]
Clear solutions are expected in the validation. In those cases where a clear solution cannot be obtained, appropriate studies should ensure that the recovery is suitable for the intended use.
Reagents: All reagents used for the preparation of sample and standard solutions should be sufficiently pure for the intended purpose.

ANALYTICAL PROCEDURES 1 AND 2

System standardization and suitability evaluation using applicable reference materials should be performed for each analytical sequence.

Procedure and Detection Technique

Procedure 1 can be used for elemental impurities generally amenable to detection by inductively coupled plasma–atomic (optical) emission spectroscopy (ICP–AES or ICP–OES). *Procedure 2* can be used for elemental impurities generally amenable to detection by inductively coupled plasma-mass spectrometry (ICP–MS). Before initial use, the analyst should verify that the procedure is appropriate for the instrument and sample used (procedural verification) by meeting the procedure validation requirements below.

Procedure 1: ICP–OES

Standard solution 1: 1.5*J* of the *Target element(s)* in a *matrix matched solution*

Standard solution 2: 0.5*J* of the *Target element(s)* in a *matrix matched solution*

Sample stock solution: Proceed as directed in *Sample Preparation* above. Allow the sample to cool, if necessary. For mercury determination, add an appropriate stabilizer.

Sample solution: Dilute the *Sample stock solution* with an appropriate solvent to obtain a final concentration of the *Target element(s)* within the calibrated range.
Blank: *Matrix matched solution*

Elemental spectrometric system

Mode: ICP

Detector: Optical detection system

Rinse: Diluent used

Standardization: *Standard solution 1*, *Standard solution 2*, and *Blank*

System suitability Sample: Standard solution of the Target element(s) in a matrix matched solution at a concentration within the calibrated range

Suitability requirements

Short term Instrumental Stability: Compare results obtained from *System suitability sample* before and after the analysis of the *Sample solution*.

Suitability criteria: NMT 20% deviation from the theoretical concentration of the system suitability sample. [NOTE—If samples are high in mineral content, rinse the system well in order to minimize carryover and check it by measuring a blank solution before introducing the *System Suitability Sample*.]

Analysis: Analyze according to the manufacturer's suggestions for program and wavelength. Calculate and report results on the basis of the original sample size. [NOTE—Appropriate measures must be taken to correct for matrix-induced interferences (e.g., wavelength overlaps).]

Procedure 2: ICP–MS

Standard solution 1: 1.5*J* of the *Target element(s)* in a *matrix matched solution*

Standard solution 2: 0.5*J* of the *Target element(s)* in a *matrix matched solution*

Sample stock solution: Proceed as directed for *Sample Preparation* above. Allow the sample to cool, if necessary. For mercury determination, add an appropriate stabilizer.

Sample solution: Dilute the *Sample stock solution* with an appropriate solvent to obtain a final concentration of the *Target element(s)* within the calibrated range.

Blank: *matrix matched solution*

Elemental spectrometric system

Mode: ICP. [NOTE—An instrument with a cooled spray chamber is recommended. (A collision cell or reaction cell may also be beneficial.)]

Detector: Mass spectrometer

Rinse: Diluent used

Standardization: *Standard solution 1*, *Standard solution 2*, and *Blank*

System suitability Sample: *Standard solution of the Target element(s) in a Matrix matched solution at a concentration within the calibrated range*

Suitability requirements

Short term Instrumental Stability: Compare results obtained from *system suitability sample* before and after the analysis of the *Sample solution*.

Suitability criteria: NMT 20% deviation from the theoretical concentration of the system suitability sample. [NOTE—If samples are high in mineral content, rinse the system well in order to minimize carryover and check it by measuring a blank solution before introducing the *System suitability sample*.]

Analysis: Analyze according to the manufacturer's suggestions for program and *m*/*z*. Calculate and report results based on the original sample size. [NOTE—Appropriate measures must be taken to correct for matrix-induced interferences (e.g., argon chloride interference with arsenic determinations).]

REQUIREMENTS FOR PROCEDURE VALIDATION

All procedures must be validated in accordance with the validation requirements described below. The level of validation necessary to ensure that a procedure is acceptable depends on whether a limit test or a quantitative determination is used. Any procedure that has been validated and meets the acceptance criteria that follow is considered to be suitable for use.

During procedure validation, the system suitability requirements as established for the procedure must be met.

PROCEDURES FOR LIMIT TESTS

The following section defines the validation parameters for the acceptability of limit tests. Meeting these requirements must be demonstrated experimentally using appropriate tests and reference material. The suitability of the method must be determined by conducting studies with the material or mixture under test spiked with known concentrations of each *Target element* of interest at the appropriate target concentration.

Detection Limit

The detection limit is shown to be sufficiently low by the analysis of samples with known concentrations of analyte at and below the target concentration.

For the purposes of this chapter, detection limit does not mean that the procedure must demonstrate lowest possible analytical result.

Standard solution: A preparation of reference materials for the *Target element(s)* at 1.0 J *in a matrix matched solution.*

Spiked sample solution 1: Prepare a solution of sample under test, spiked with appropriate reference materials for the *Target element(s)* at the *Target concentration*, solubilized or digested as described in *Sample Preparation*.

Spiked sample solution 2: Prepare a solution of the sample under test, spiked with appropriate reference materials for the *Target element(s)* at 80% of the *Target concentration*, solubilized or digested as described in *Sample Preparation*.

Unspiked sample solution: A sample of material under test, solubilized or digested in the same manner as the spiked *sample solutions.*

Acceptance criteria

Non-instrumental procedures: *Spiked sample solution 1* provides a signal/response, e.g., color, or intensity equivalent to or greater than that of the *Standard solution*. *Spiked sample solution 2* must provide a signal /response, e.g., color, or intensity less than that of *Spiked sample solution 1*. [NOTE—The signal/response, e.g., color, or intensity from each *Spiked sample solution* is NLT the *Unspiked sample solution* determination.]

Instrumental procedures: The average value of the three replicate measurements of *Spiked sample solution 1* is within ±15% of the average value obtained for the replicate measurements of the *Standard solution*. The average value of the replicate measurements of *Spiked sample solution 2* must provide a signal intensity or value less than that of the *Standard solution*. [NOTE—Correct the values obtained for each of the spiked solutions using the *Unspiked sample solution*.]

Specificity

The procedure must be able to unequivocally assess each *Target element* in the presence of components that may be expected to be present, including other *Target elements*, and matrix components.

Precision, only for Instrumental Methods (Repeatability)

Sample solutions: Six independent samples of the material under test, spiked with appropriate reference materials for the *Target element(s)* at the *Target concentration*.

Acceptance criteria

Relative standard deviation: NMT 20% for each *Target element*

PROCEDURES FOR QUANTITATIVE TESTS

The following section defines the validation parameters for the acceptability of procedures for quantitative tests. Meeting these requirements must be demonstrated experimentally, using appropriate tests and reference materials.

Accuracy

Standard solutions: Prepare solutions containing the *Target element(s)* at three concentrations ranging from 0.5 *J* to 1.5 *J*, using appropriate reference materials, in a Matrix matched solution.

Test samples: Prepare 3 independent sample preparations of the material under test spiked with appropriate reference materials for the *Target element(s)* at the target concentration,J, before any sample preparation steps (digestion or solubilization). Spike concentrations should range from 0.5 *J* to 1.5 *J* and should include at least 3 individual concentrations.

Acceptance criteria

Spike recovery: 70%–150% for the mean of three independent sample preparations at each concentration

Precision

REPEATABILITY

Test samples: Six independent samples of material under test (taken from the same lot) spiked with appropriate reference materials for the *Target element(s)* at the Target concentration or at least 9 determinations (e.g. 3 replicates of 3 concentrations) covering the specified range.

Acceptance criteria

Relative standard deviation: in both cases, NMT 20% for each *Target element*

INTERMEDIATE PRECISION (RUGGEDNESS)

Perform the *Repeatability* analysis again, either on a different day, with a different instrumentation, with a different analyst, or a combination thereof. Combine the results of this analysis with the *Repeatability* analysis.

Acceptance criteria

Relative standard deviation: NMT 25% for each *Target element*

Specificity

The procedure must be able to unequivocally assess each *Target element* in the presence of components that may be expected to be present, including other *Target elements*, and matrix components.

Range and Linearity

Demonstrated by meeting the *Accuracy* requirement.

Quantitation Limit

Use the results from the accuracy study.

QL of 0.5 J is confirmed when the accuracy acceptance criteria for 0.5 J spiked solution is met.

Acceptance criterion: the QL is less than or equal to 0.5 J.

GLOSSARY

Concentrated acid: Concentrated ultra-pure nitric, sulfuric, hydrochloric, hydrofluoric acids or any other acid or mixture of acids that is demonstrated to be suitable.

Matrix matched solution: Solutions having the same solvent composition as the *Sample solution*. In the case of an aqueous solution, *Matrix matched solution* would indicate that the same acids, acid concentrations, and mercury stabilizer are used in both preparations.

Target elements: Elements which must be evaluated according to the requirements defined in other chapters.

Target limit or Target concentration: The acceptance value for the elemental impurity being evaluated. Exceeding the *Target limit* indicates that a material under test exceeds the acceptable value. [NOTE—*Target limits* can be approximated by dividing the *permitted daily exposures (PDEs)* by the maximum daily dose of the drug product.].]

J: Final concentration of the Target element(s) in the standard and the sample solutions. It corresponds to the concentration (w/v) of the Target element(s) at the *Target limit*, appropriately diluted to the working range of the instrument. If a dilution is not necessary J is equal to the target concentration. For example, if the target elements are lead and arsenic for an analysis of an oral solid drug product with a daily dose of 10 g/day using inductively coupled plasma–mass spectrometry (ICP–MS), the target limit for these elements would be 0.5 µg/g and 1.5 µg/g. However, in both cases, the linear dynamic range of the ICP–MS is known to extend from 0.01 ng/mL to 0.1 µg/mL for these elements. Therefore, a dilution factor of at least 1:100 is required to ensure that the analysis occurs in the linear dynamic range of the instrument. *J* would thus equal 5 ng/ml and 15 ng/mL for lead and arsenic, respectively (Note: the density of the sample solution may have to be considered).

Appropriate reference materials: Where *Appropriate reference materials* are specified in the chapter, certified reference materials (CRM) from a national metrology institute (NMI), or reference materials that are traceable to the CRM of an NMI should be used.